# **Exploration of Plasmonic Copper-Based Nanocrystals Through Various Synthetic Methods** Jonathan B. Rider, Jason E. Kuszynski, Catherine J. Fabiano, Geoffrey F. Strouse Department of Chemistry and Biochemistry, Florida State University





## Introduction

Nanotechnology is a young field of research with applications in electronics, medicine, and energy. Plasmonic nanomaterials scatter and manipulate light in unique ways which make them apt candidates for improved sensors, catalysis, and numerous biotechnical devices. The study centered around two techniques- microwave reactor and roundbottom synthesis. The microwave reactor method involves irradiating the sample with microwave radiation at continuous wattage and programmed temperature, while round bottom uses a traditional, convection heat source with a thermocouple. The particles are cooled, rid of unwanted material, and vacuumed to yield solid nanocrystals. Spectroscopic techniques are then used to reveal the structural and energetic characteristics of the material. X-ray diffraction (XRD) identifies crystalline phases and serves as the initial indicator of synthetic success. Ultraviolet-Visible (UV-VIS) extinction spectroscopy then tests for the presence of a localized surface plasmon. If a plasmon is present, a large, broad extinction is observed.

### Methods

### **Chemical Species:**

- Copper (I) Acetate, Cu(OAc), 97%
- Copper (II) Acetylacetonate, Cu(acac)<sub>2</sub>, 97%
- Chromium (III) Acetylacetonate, Cr(acac)<sub>3</sub>, 97%
- -1-octadecene, ODE, 90%
- -1-dodecanethiol, DDT, 98%
- Oleylamine, OLAM, 70%

### **Round-Bottom Synthesis**

- 1) Cu(OAc) and  $Cr(acac)_3$  are added to a 50 mL round bottom flask with ODE at room temperature
- 2) DDT and OLAM are prepared as an injection in a separate vial
- 3) The solution is heated to 280°C under a nitrogen atmosphere
- 4) Injection occurs, flask contents are maintained at 280°C for 30 min
- 5) Particles are cleaned with strong solvents and centrifuged for 10 min **Microwave Reactor Synthesis**
- 1) Cu(OAc) and  $Cr(acac)_3$  are added to a microwave vial with ODE at room temperature
- 2a) For one-vial synthesis, DDT and OLAM are also added to the vial
- 2b) For injection synthesis, step 2 is completed as described above
- 3) The solution is placed in the microwave and irradiated with continuous power of 300 W for 10 min
- For injection synthesis, injection occurs after the first 5 min 4) Particles are cleaned with strong solvents and centrifuged for 10 min



Figure I: Round-Bottom Flask Reactor Middle Neck: Condenser Right Neck: Thermocouple

# Setup



Figure III: Cu(acac)<sub>2</sub>, DDT, Cr(acac)<sub>3</sub> Microwave Vial

### Results



Figure II: CEM Discover Microwave Synthesis Reactor



Figure IV: Top: 3 mL Sulfur/OLAM Injection Bottom: 1 mL DDT Injection

The round bottom synthesis method was more effective than microwave methods in yielding nanocrystals with localized surface plasmons. XRD data confirmed both methods produced the desired Copper-based product, but preliminary UV-VIS data could only identify extinction characteristic of plasmonic behavior in the crystals made through the round bottom technique.

Changing the initial one-vial microwave technique to include a continuous injection drip appeared to help the reaction proceed to completion. Furthermore, heating the solid components of the reaction into solution before starting injection minimized the release of unwanted gaseous products like Sulfur.

Future work will be dedicated to studying different microwave methods and evaluating alternative copper precursors with the hope of consistently observing plasmonic behavior and tuning particle size.



1. Colloidal Synthesis of Cu-M-S (M = V, Cr, Mn) Nanocrystals by Tuning the Copper Precursor Reactivity Valeria Mantella, Seyedeh Behnaz Varandili, James R. Pankhurst, and Raffaella Buonsanti Chemistry of Materials 2020 32 (22), 9780-9786

2. Synthesis and Size-Dependent Optical Properties of Intermediate Band Gap Cu<sub>3</sub>VS<sub>4</sub> Nanocrystals Valeria Mantella, Silviya Ninova, Servio Saris, Anna Loiudice, Ulrich Aschauer, and Raffaella Buonsanti Chemistry of Materials 2019 31 (2), 532-540 DOI: 10.1021/acs.chemmater.8b04610





### Discussion

### References

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