

### Introduction

Quantum information science is gaining huge attention recently because it has the potential to significantly advance various fields of human activity. One crucial factor in this is the phase memory time  $(T_2)$ , which needs to be extended in order to make computing algorithms practical. Our project focuses on uncovering and studying the basic magnetic properties of certain coordinated complexes. Coordinated complexes are molecular groups enveloping a central metal.

Specifically, we are searching for systems where the energy gap between avoided energy levels, known as clock transitions, is precisely 94 GHz - a frequency that our pulse instrument can access. To identify potential candidates for these clock transitions, we began with a literature review aided by ChatGPT. We then gathered spin Hamiltonian parameters from various scientific articles. Hamiltonian parameters provide details that are needed to understand the energy and interaction within the system. Using Matlab simulation in collaboration with the EasySpin package, we created energy level plots to assess these candidates. We then extracted the promising candidates and tested them using High Frequency Electron Paramagnetic Resonance (HFEPR) at Maglab to assess their realworld behavior and validate the theoretical predictions generated through our simulations. By subjecting our selected candidates to the controlled conditions provided by Maglab we aim to gain empirical evidence that support and refines our theoretical understanding, ultimately contributing to the broader exploration of quantum information science.



# Clock Transitions for Quantum Computing Olivia Jernstedt and Dr. Jakub Hrubý Florida State University, Tallahassee, Fl

### Materials/Methods

### Materials

- 99.999% Nickel (II) Sulfate Heptahydrate
- 99.95% Zinc (II) Sulfate Heptahydrate
- 98% 4-Pentyl-4-Biphenylcarbonitrile
- Fisher Stirring Hotplate
- **Stock Solutions**
- $0.5607 \text{ g of NiSO}_{4} \cdot 7H_{2}O + 20 \text{ mL } H_{2}O$
- 1.4011 g of  $ZnSO_4 \cdot 7H_2O + 20 mL H_2O$
- **Crystal Solutions**
- 10%
- 5 mL of NiSO<sub>4</sub>•7H<sub>2</sub>O mixed with 5 mL of  $ZnSO_4$ •7H<sub>2</sub>O + 5 mL H<sub>2</sub>O 1%
- 0.5 mL of NiSO<sub>4</sub>•7H<sub>2</sub>O mixed with 5 mL of  $ZnSO_4$ •7H<sub>2</sub>O + 5 mL H<sub>2</sub>O 0.1% •
- 0.05 mL of NiSO<sub>4</sub>•7H<sub>2</sub>O mixed with 5 mL of  $ZnSO_4$ •7H<sub>2</sub>O + 5 mL H<sub>2</sub>O 1% with liquid crystal
- 0.5 mL NiSO<sub>4</sub>•7H<sub>2</sub>O mixed with 5 mL ZnSO<sub>4</sub>•7H<sub>2</sub>O + 5 mL H<sub>2</sub>O, + 0.25 mL of Liquid Crystal
- $NiSO_4 \cdot 7H_2O$  with liquid crystal
- $15 \text{ mL NiSO}_4 \cdot 7H_2O + 0.25 \text{ mL of Liquid Crystal}$

## Results

The experiment is currently ongoing. We hope to see that at least one of these components displays a clock transition within measurable range.



• Matlab and EasySpin's simulation of  $Ni(H_2O)_6^{2+}$  in  $NiSO_4 \cdot 7H_2O$ 



# • $S = 1, D = -3.5 \text{ cm}^{-1}, E = -1.5 \text{ cm}^{-1}, g = 2.2$

- 309.
- Spectroscopy, 128, 342–350.
- hyperfine interaction.

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### Acknowledgment

Growing phase of crystal compounds